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Steve Zumda hl is the author of market-leading textbooks in introductory chemistry, general chemistry, honors-level chemistry, and high school chemistry. Recently retired from his long-time position as Director of Undergraduate Programs at the University of Illinois, he has received numerous awards for his contributions to chemical education.

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Run the same experiment on a Sunday, when most of the general population doesn't work and when a significant number of downtown stores are closed in the morning. 123. Let x = mass of copper and y = mass of silver. + x y 105.0 g = x + y and 10.12 mL = ; solving and carrying 1 extra sig.

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Also, since 1 cm³ # 1 milliliter (mL), 1 L 5 1000 cm³ 5 1000 mL Thus 1 liter contains 1000 cubic centimeters, or 1000 milliliters. Chemical laboratory work frequently requires measurement of the volumes of liquids. Several devices for the accurate determination of liquid volume are shown in Fig.

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